

Chlorination and Arsenic Treatment

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Utah Division of Drinking Water
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Presentation Topics

- 3 Types of Chlorination
 1. Chlorine (Gas)
 2. Sodium Hypochlorite (Liquid)
 - Injection/Dilution
 - On-site Generation
 3. Calcium Hypochlorite (Tablet/Granule/Powder)
- Arsenic
- Arsenic Compliance Strategies



Different Types of Chlorine

Gas - Chlorine



Bleach - Sodium Hypochlorite



Powder/Tablets - Calcium Hypochlorite



☞ All three forms of chlorine will react with water to form *hypochlorous acid (HOCl)*



Disinfection Rule (R309-520)

- Chlorine (R309-520-7)
 - Gaseous chlorine
 - Calcium hypochlorite
 - Sodium hypochlorite liquid
 - Sodium hypochlorite on-site generation
- UV
- Ozone
- Chlorine Dioxide
- Chloramines



Disinfection Rule - General

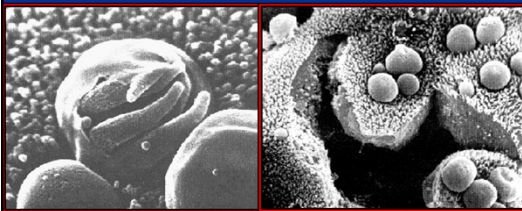
- Primary Disinfectant
 - Chlorine, Ozone, UV, Chlorine Dioxide
 - Treatment process
 - CT & UV dose
- Secondary Disinfectant
 - Chlorine & Chloramines
 - Maintain residual in distribution system
- Continuous Disinfection (Surface water & GWUDI)
- ANSI/NSF Standard 60 certified chemicals



Disinfection Rule - General

■ Bin 1 Surface Water

2-log *Cryptosporidium*



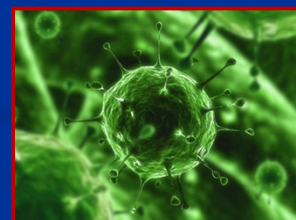
3-log *Giardia*



E. Coli Monitoring



4-log virus



Chlorine Rule — General

- Automatic proportioning
- Minimization of chlorinated overflow
- Chlorination facility design
 - Heating, lighting, ventilation
 - Feed water piping, flow measurement
 - Standby and backup equipment
 - Bypass-to-waste and isolation
 - Chlorinator capacity



Chlorinator Design & Required CT

- CT → Concentration and Time
- 4-log virus inactivation → CT=12
- High quality GW source
 - CT of 12 not required when adding new chlorinator
 - Recommended
- Questionable GW source
 - CT of 12 is required when adding new chlorinator
 - May need additional storage or special configuration
- Booster chlorinator for distribution system
 - CT of 12 not required
 - Maintain disinfection residual

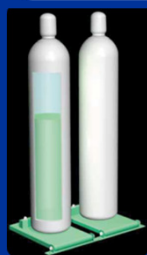
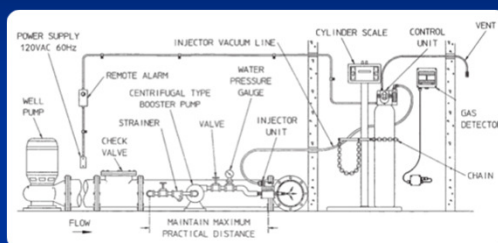


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Gas Chlorination



Gas Chlorine

Additional Requirements in R309-520-7(2)

- Automatic switch over
- Heating and ventilation
- Injector and eductor
- Housing
- Cylinder security
- Weighing scales
- Pressure gauges



Gas Chlorination

- Low chemical cost
- Low maintenance cost
- Shipped and stored in same container
- Easy to regulate / measure feed rates
- Chlorine concentration constant



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Sodium Hypochlorite Chlorination



Sodium Hypochlorite Chlorination

R309-520-7(3)(a)

- ANSI/NSF 60 chemicals
- Emergency eyewash/shower
- Minimize decay



Sodium Hypochlorite Chlorination

- Readily available chemical
- 5-15% Strength
- Relatively non-lethal
- Generally inexpensive feed equipment



Sodium Hypochlorite Chlorination Pumps

Solenoid



Peristaltic



Stepper Motor



Mechanically Actuated



Sodium Hypochlorite Chlorination Dilution

- Dilute to Any Strength
- Minimal Space Requirements
- Pre-Engineered Systems



Sodium Hypochlorite Chlorination Dilution



#1 ALARM TYPE
PUMP VAPOR LOCK

#1 ALARM TYPE
CHLORATE WARNING

#1 ALARM TYPE
HIGH LEVEL

#1 ALARM TYPE
LOW LEVEL



HYP0 = 266.5 LB
STRENGTH 12.7 %

HYP0 = 266.5 LB
100%CL2 48.0 LB

HYP0 = 266.5 LB
LB-HR= 10.2

HYP0 = 266.5 LB
DAYS OLD 47



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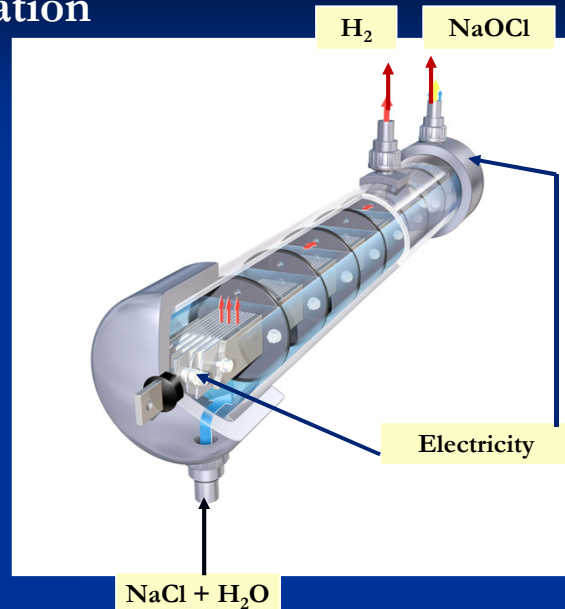


Sodium Hypochlorite Chlorination On-site Generation

R309-520-7(3)(b)

- ANSI/NSF 61
- ANSI/NSF 60

✓ Safety



Sodium Hypochlorite Chlorination On-Site Generation



Sodium Hypochlorite Chlorination On-Site Generation

- 0.8% Strength
- Variety of generation capacities
- Flexible installation
- Visual indication of cell operation



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Calcium Hypochlorite Chlorination Powder/Tablet Feeders



Calcium Hypochlorite Chlorination R309-520-7(3)(c) and (d)

- Calcium Hypochlorite
 - Tablet, Granule, Powder
 - ANSI/NSF 60 chemicals
 - Storage & Safety
 - Operation & Maintenance
- Hypochlorite Feed Equipment
 - ANSI/NSF 61 materials
 - Exception



Calcium Hypochlorite Chlorination

- Long shelf life: (~5% loss/year)
- Minimal chlorate formation
- Easy to transport and store
- Spills/leak less likely
- NSF Standard 60 Certified



Which to Choose

- Not dictated by the State
- Operator's Experience & Skill
- Cost
- Ease of Handling & Maintenance
- Safety
- Foot Print
- Existing Facility
- Power Availability
- Location

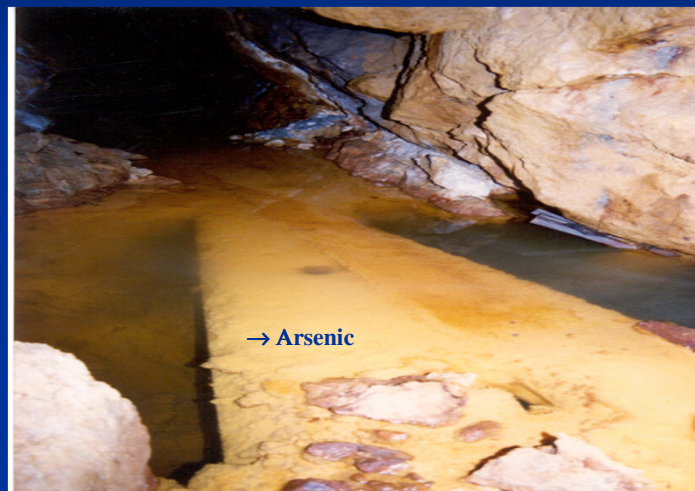


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Arsenic



Arsenic

- Maximum Contaminant Level (MCL) at 10.0 ug/L (10.0 ppb)
 - MCL based on chronic health effects
 - Multiple oxidation states
 - Arsenite (III)
 - Arsenate (V)
- $$\text{As}^{3+} \rightarrow \text{As}^{5+} + 2e^{-}$$

Arsenic Speciation

- Theoretically, As (V) in surface water & As (III) in ground water.
- Oxidize As (III) to As (V) before applying treatment.
 - As (III) has higher toxicity
 - As (V) is more effectively removed by most treatment technologies

Oxidize As (III) to As (V)

- Effective Oxidants
 - ✓ Free Chlorine
 - ✓ Potassium Permanganate
 - ✓ Ozone
- Ineffective
 - ✓ Chloramines, Chlorine Dioxide, UV, Aeration
- Point of oxidant application is important

How Are Utah Water Systems Affected?

- 37 Utah water systems' water sources exceed arsenic MCL (10.0 ug/L)
- 16 Counties affected
 - Millard (7 water systems)
 - Salt Lake (6)
 - Box Elder (4)
 - Wayne (3)

EPA Decision Tree Overview

- Step 1: **Water Quality Monitoring**
- Step 2: **Blending**
- Step 3: **Optimizing Existing Treatment**
 - Enhanced Coagulation/Filtration
 - Enhanced Lime Softening
 - Iron & Manganese Filtration
- Step 4: **Selecting New Treatment**
 - Ion Exchange Processes
 - Adsorption Processes
 - Membrane Processes

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- **Arsenic Compliance Strategies**



Arsenic Compliance Options

1. Abandon source or develop alternate source
2. Sample averaging
3. Blend with other sources
4. Combination of blending & averaging
5. Point of use/point of entry device
6. Central treatment

Blend with Other Sources

- Identify flow and arsenic level of possible sources
- Compare blended concentration with MCL
- Evaluate system layout, blending mechanism and sampling location (before 1st customer)
- Construction or SCADA may be needed
- Blending plan must be reviewed and approved

Point of Use (POU)



POU/POE for Arsenic Removal

- All homeowners agree (100% participation)
 - Public meeting on costs and options
 - If POU is the majority position, a POU unit installed at each kitchen tap & water user allows PWS the access for servicing of units
- Only some taps treated
- Rotating sampling sites within the PWS
- O&M and replacement
- PWS sends operation reports to DDW
- Using POU/POE for arsenic compliance must be reviewed and approved

Point of Use Technologies



- Ion exchange
- Activated alumina
- Granular ferric hydroxide
- Reverse Osmosis (RO)

Reverse Osmosis POU



Arsenic Compliance Options

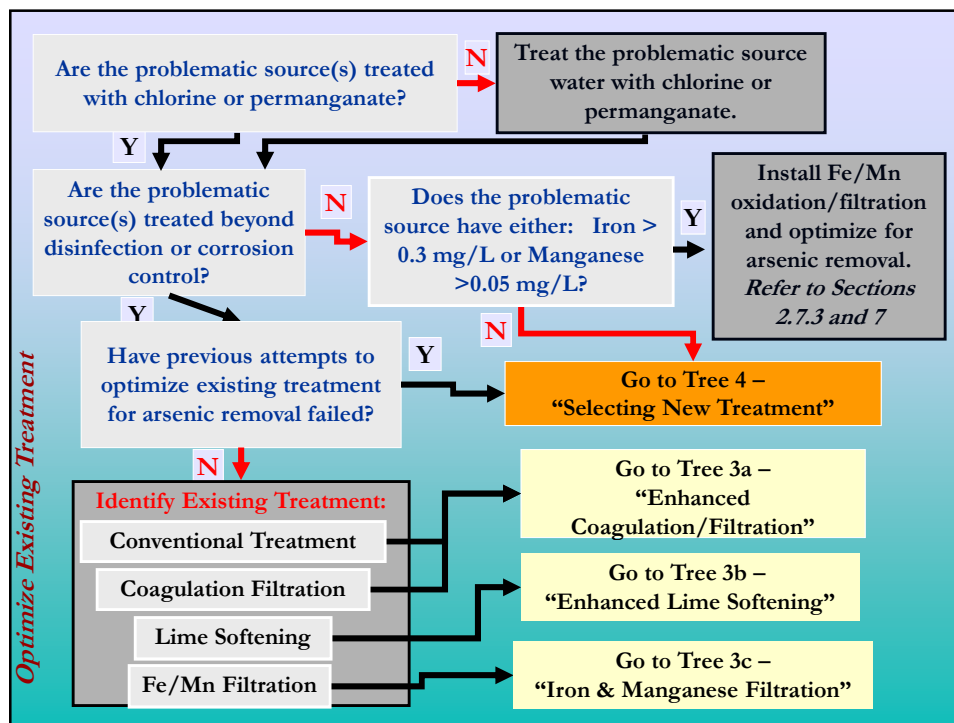
1. Abandon source or develop alternate source
2. Sample averaging
3. Blend with other sources
4. Combination of blend & averaging
5. Point of use/point of entry device
6. **Central treatment**

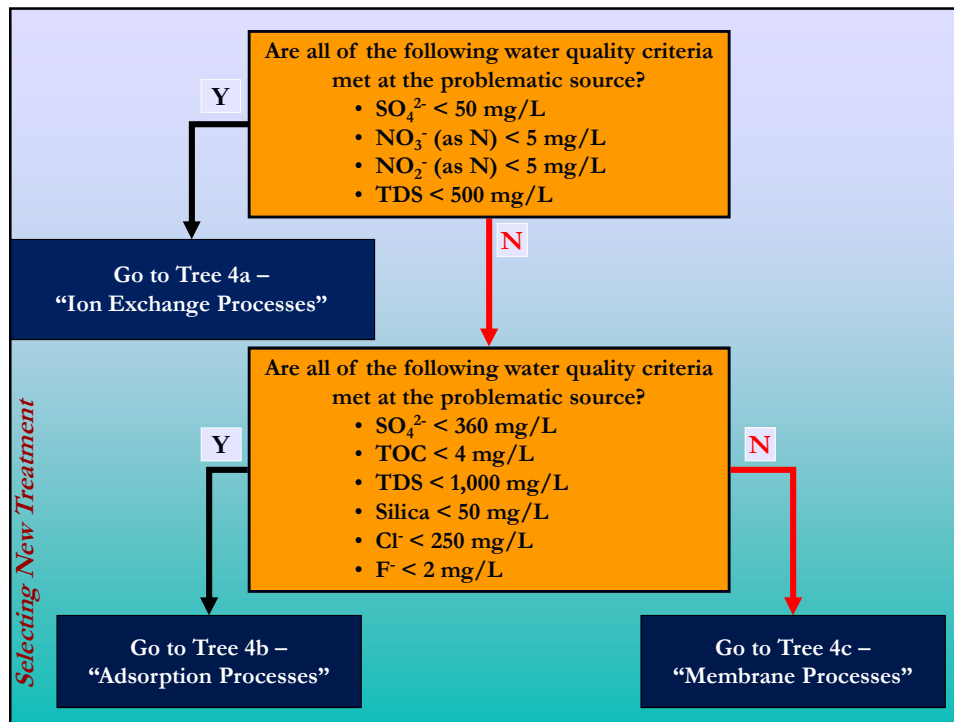
Arsenic Mitigation in Utah

1. Abandon source or develop alternate source **(4 water systems)**
2. Sample averaging **(2)**
3. Blend with other sources **(9)**
4. Combination of blend & averaging **(2)**
5. POU **(8)** & POE **(0)**
6. Central treatment **(10)**
7. No longer PWS **(1)**

EPA Decision Tree Overview

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 - Sorption Processes
 - Membrane Processes





Optimization of Existing Process

1. Enhanced Coagulation/Filtration
2. Enhanced Lime Softening
3. Iron and Manganese Removal
 - Oxidation/Filtration

Optimization of Existing Process

To Increase Arsenic Removal

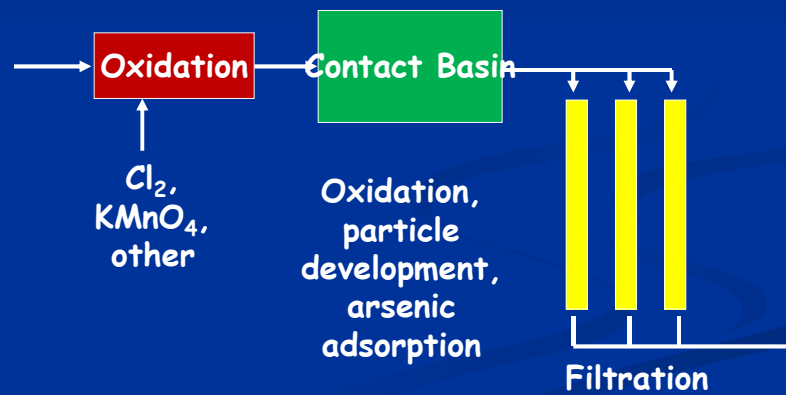
Utility with existing iron removal process
and not meeting 10.0 ppb:

1. Increase iron concentration
2. Adjust pH
3. Replace arsenic adsorption media
4. Change point of oxidant addition

Iron-Based Arsenic Removal Process

- Adsorptive properties of iron mineral toward arsenic
- Arsenic treatment processes:
 - ✓ Iron Removal
 - ✓ Coagulation with iron coagulant
 - ✓ Iron-based adsorption media

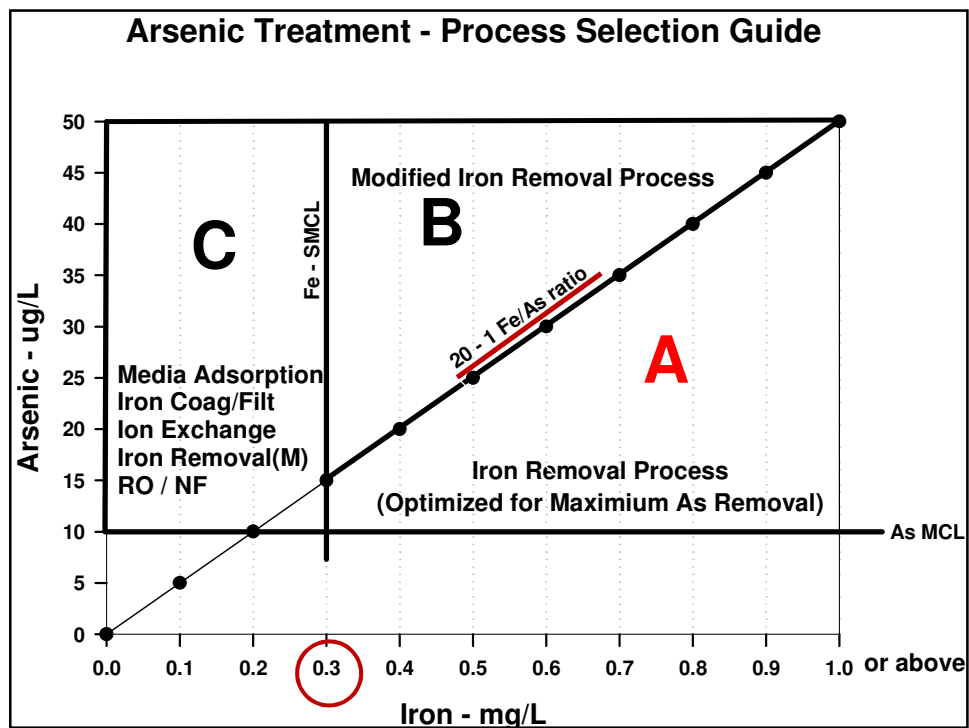
Iron and Arsenic (& Mn) Removal



Iron-Based Arsenic Removal Process

Removal of **1** mg/L of iron
achieves
removal of **50** ug/L arsenic

* Optimized conditions and 100% As (V)



Installation of New Treatment

1. Ion Exchange
2. Membrane
3. Adsorption

Raw Water Testing

■ Primary parameters

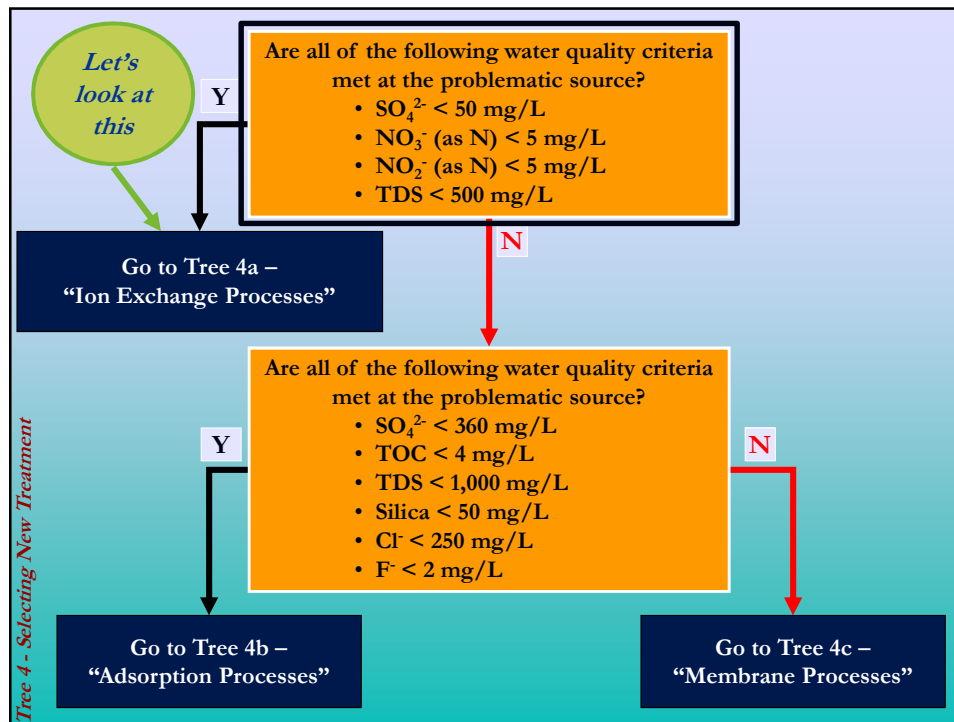
- Total Arsenic, Arsenite, Arsenate
- Chloride
- Fluoride
- Iron
- Magnesium
- Manganese
- Nitrate/Nitrite
- Orthophosphate
- pH
- Silica
- Sulfate
- Total Dissolved Solids (TDS)

■ Secondary parameters

- Alkalinity
- Aluminum
- Calcium
- Turbidity
- Hardness

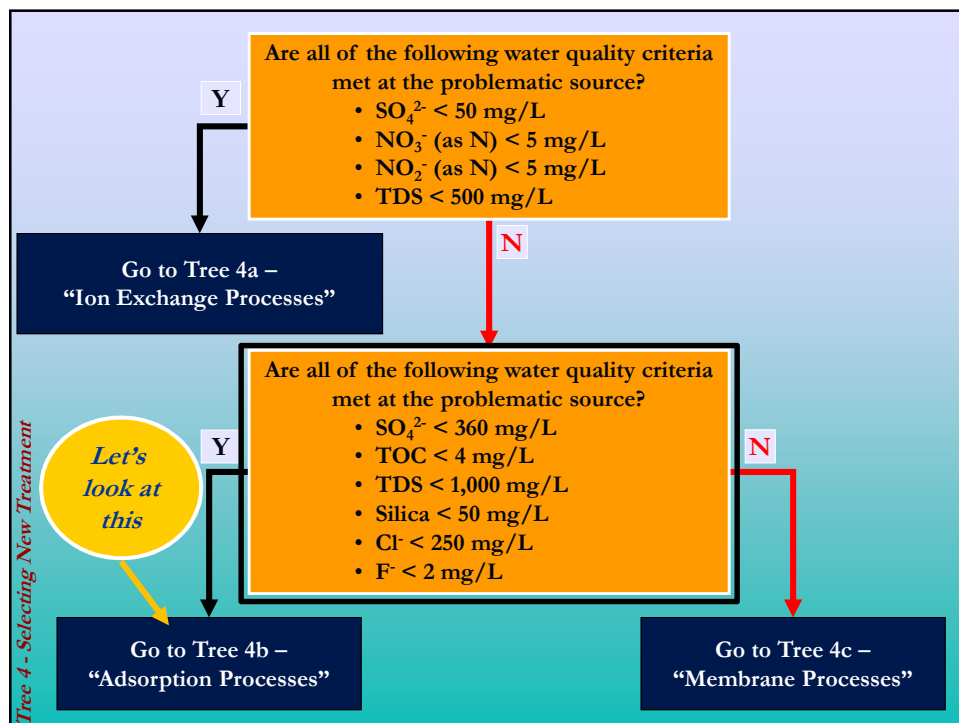
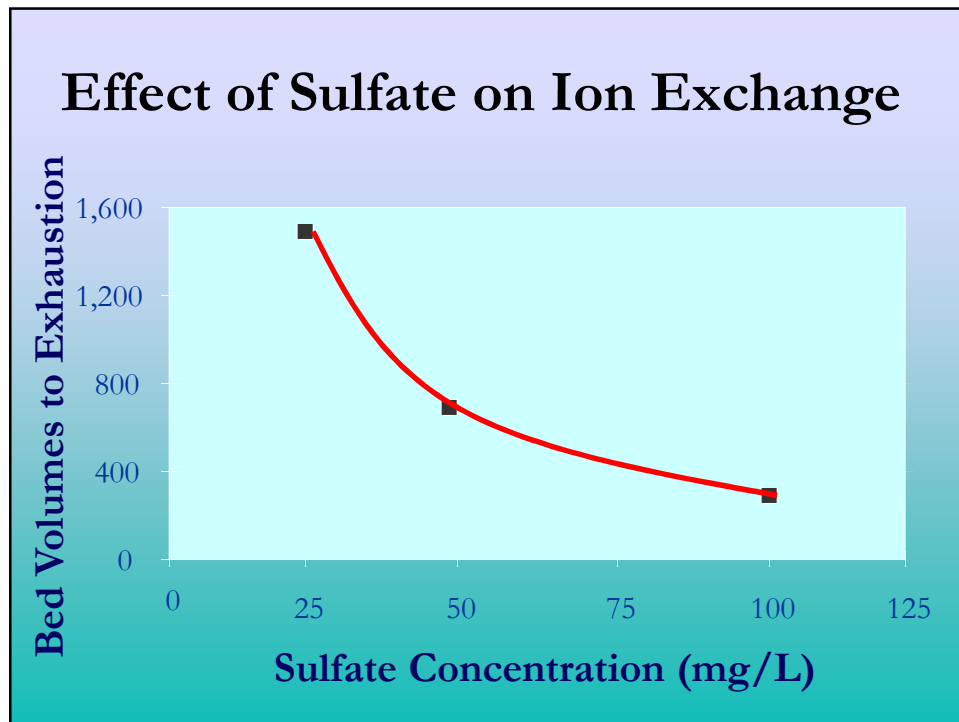
Arsenic Treatment & Water Quality

<u>Parameter</u>	<u>St. George</u>	<u>Magna</u>
Total Arsenic	24.5 ppb	11.3 ppb
Iron	0.1 mg/L	0.069 mg/L
Manganese	<0.01 mg/L	0.066 mg/L
Total Silica	21.5 mg/L	70 mg/L



Ion Exchange for Arsenic Removal

- Ions exchanged between a solution phase and solid resin phase
- Exchange affinity
 - Sulfate, TDS, selenium, fluoride, nitrate
- Insensitive to pH
- Consider waste by-product disposal
- Suitable for small GW systems with low sulfate and TDS and as the polishing step after filtration



Adsorption Processes

1. Activated Alumina (AA)
2. Granular Ferric Hydroxide (GFH)



Activated Alumina

Delta
City

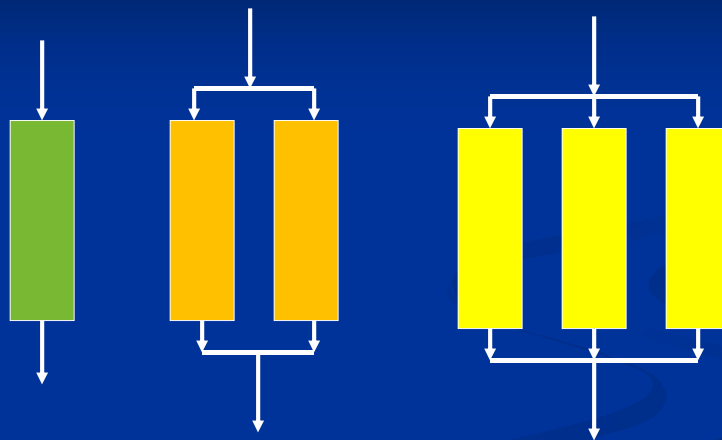


Granular Ferric Hydroxide

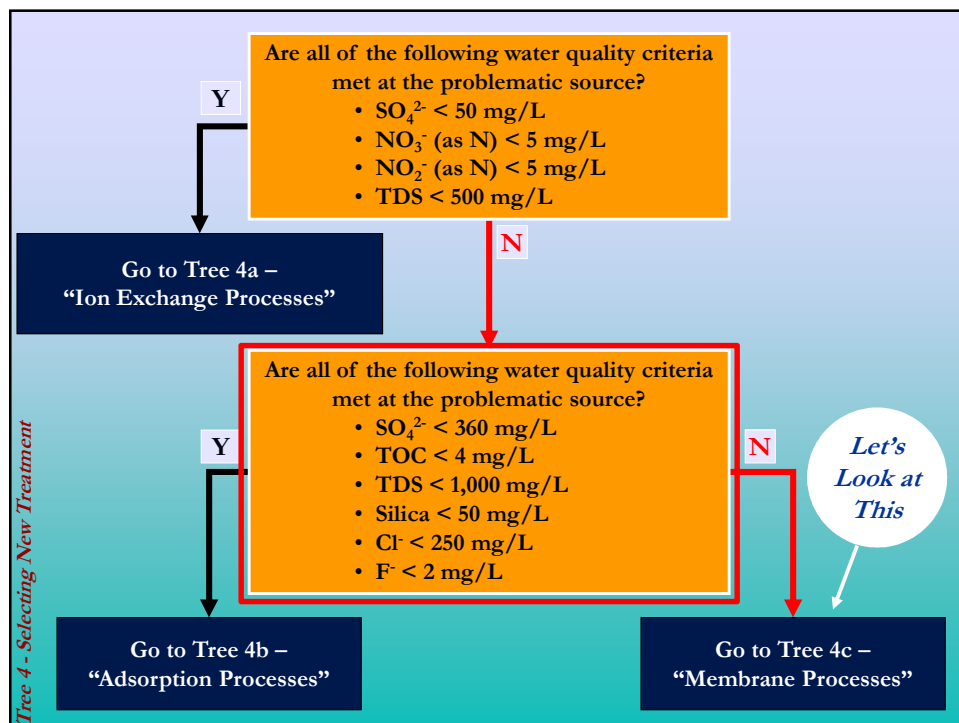
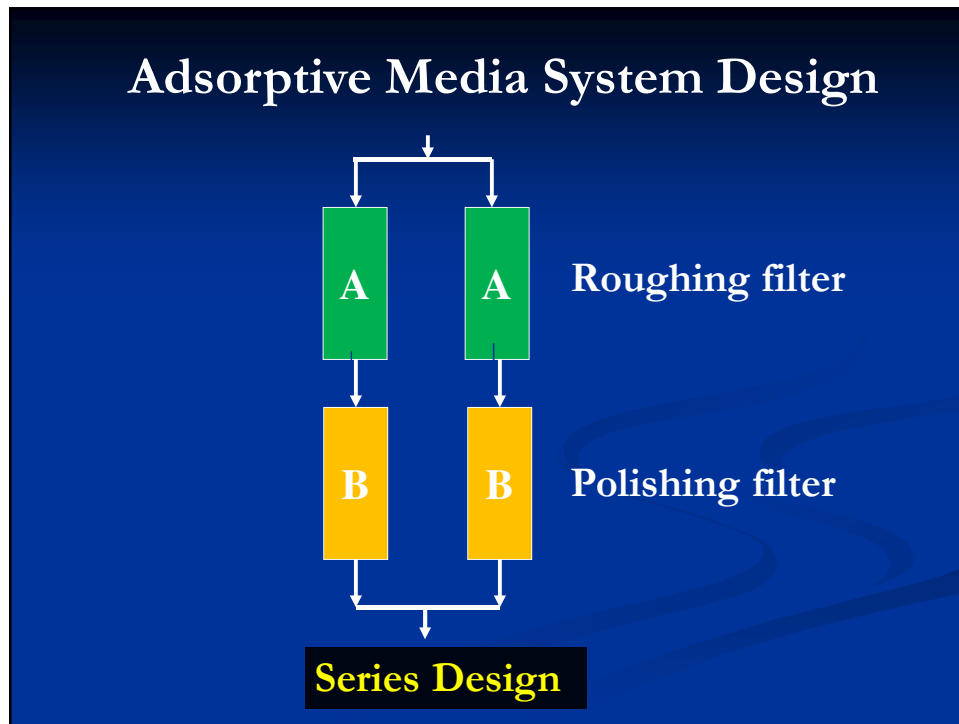
Activated Alumina System



Adsorptive Media System Design



Simple 1, 2, or 3 beds in parallel



Membrane Processes

- RO
- Nanofiltration
- Coagulation Assisted Microfiltration

Conclusions

- Treatment is only one of the many mitigation options
- Iron removal → arsenic removal
- Arsenic speciation and oxidation are important
- Water quality affects selection & performance of arsenic treatment

Questions?

Utah Division of Drinking Water

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